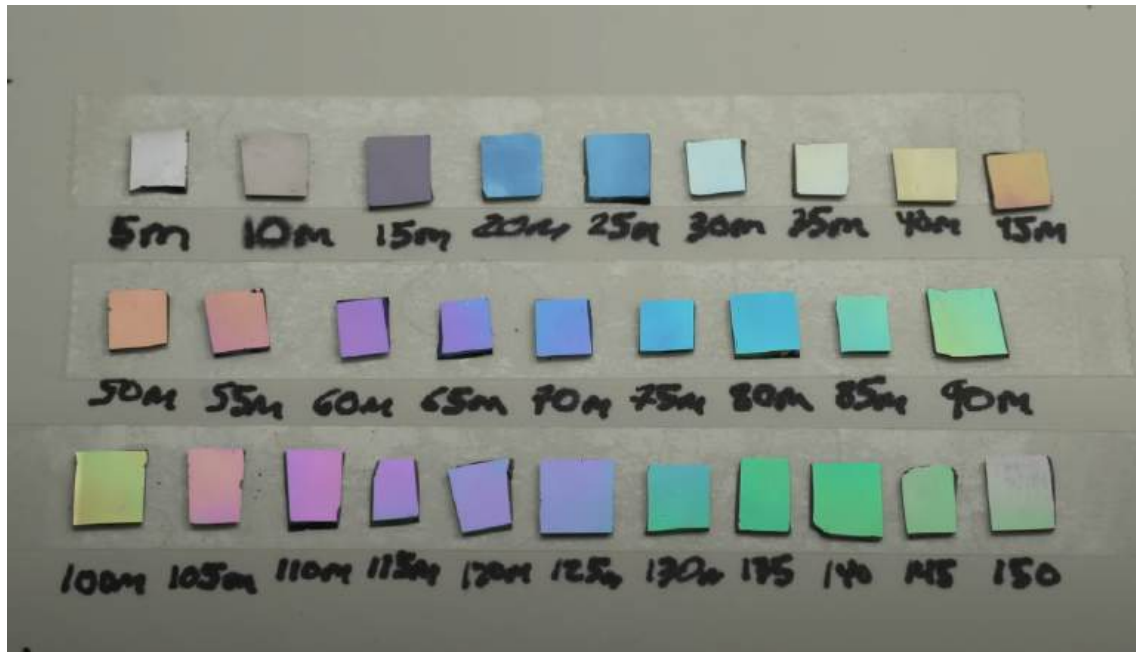


T1. Thermal oxidation

Thermal oxidation is the process of creating a very thin (tens/hundreds of nanometres) film of silicon dioxide SiO_2 on the surface of silicon Si at high temperatures. This technology is an essential part of the manufacturing process of integrated circuits and MOSFETs (metal-oxide-semiconductor), which form the basis of modern electronics. If you place silicon Si in an aerobic environment, the oxidation process of $\text{Si} + \text{O}_2 \rightarrow \text{SiO}_2$ will begin. For the description of crystalline silicon and flat oxide films of sufficient thickness (starting from ten nanometres), the Dill-Grove model (1965) does well, but a detailed description of the growth mechanism in other conditions is still an unsolved scientific problem.

Silicon is opaque to visible light, while its dioxide is transparent. These optical properties, combined with the fact that the dioxide film has a thickness of tens of nanometres, create ideal conditions for very precise optical control of the thickness of the SiO_2 layer.



The change in the visible color of the silicon substrate at different oxidation times.

"Growing Colorful Oxide Layers on Silicon" by ProjectsInFlight on YouTube.

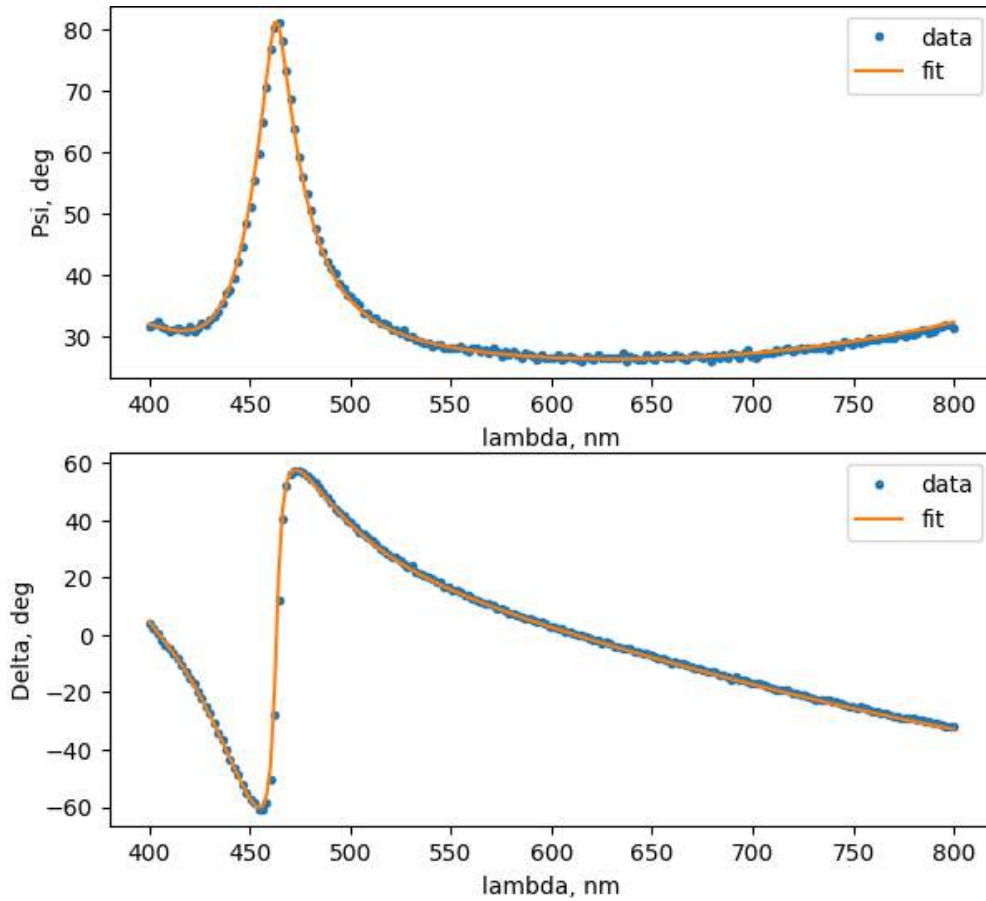
We will investigate the film growth by using ellipsometry, a powerful technique for studying the optical properties of thin films. It is based on a comparison of the amplitude coefficients of reflection of s -polarized and p -polarized waves (from the German "senkrecht" – perpendicular, "parallel" – parallel) from the surface of the film. The measurement results are represented by ellipsometric angles Ψ , Δ :

$$\frac{r_p}{r_s} = \tan \Psi \cdot e^{i\Delta},$$

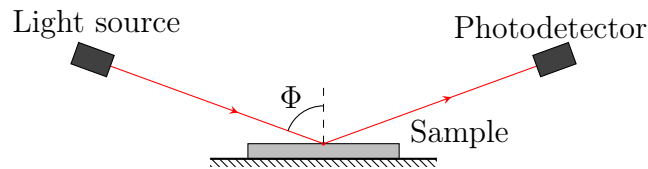
where the value of Ψ corresponds to the ratio of amplitudes upon reflection of p - and s -polarized waves, and the value of Δ corresponds to the phase difference between them.

Ellipsometry measures precisely the relative difference between waves (ratio of amplitudes and phase difference) with different polarizations, so it is very resistant to various fluctuations in the power of the radiation source and the sensitivity of the receiver. Ellipsometry also makes it possible to measure the characteristics of a sample without direct mechanical contact with it, which allows measurements to be carried out in real time and *in situ* (from lat. – "on the spot", directly).

During the entire task, the wavelength of the incident wave is marked as λ . In the optical range, substances do not have significant magnetic qualities, therefore, all materials under consideration have $\mu = 1$.



Example of an ellipsometric spectrum Ψ, Δ



Optical scheme of the ellipsometer: a light source, a Photodetector, and a test sample.

Attention! Place the files you want to attach to the report in the “report” folder.

Attention! It is enough to make only **A7** to moving on to part B.

Part A. Theoretical principles of ellipsometry

Maxwell’s equations in a non-magnetic dielectric have the form

$$\begin{aligned} \operatorname{div} \vec{D} &= 0 & \operatorname{rot} \vec{E} &= -\frac{\partial \vec{B}}{\partial t} \\ \operatorname{div} \vec{B} &= 0 & \operatorname{rot} \vec{B} &= \mu_0 \frac{\partial \vec{D}}{\partial t} \end{aligned}$$

For a dielectric:

$$\vec{D} = \varepsilon_0 \varepsilon \vec{E}$$

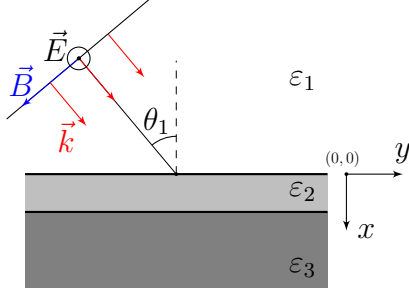
It follows from these equations that the normal components of the vectors $\vec{D}$ and $\vec{B}$ as well as the components of the vectors $\vec{E}$ and $\vec{B}$ directed along the surface do not change at the dielectric-dielectric boundary.

In a plane wave the wave vector, the electric field vector and the magnetic field vector $\vec{k}$, $\vec{E}$ and $\vec{B}$ are pairwise perpendicular and right-handed orientated. Also hold the following relations:

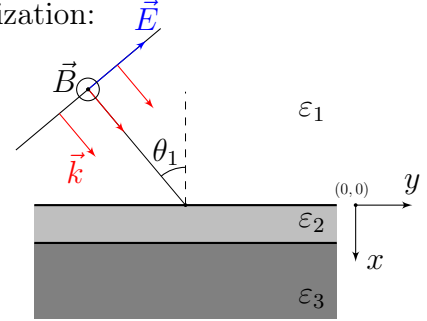
$$\vec{B} = \frac{1}{\omega} [\vec{k} \times \vec{E}], \quad |\vec{k}| = \sqrt{\varepsilon} \omega / c$$

where ω is the frequency of the wave. The speed of light $c = 1/\sqrt{\varepsilon_0\mu_0}$. As a part of this task, we will consider the interaction of a plane wave with layered structures and develop a general approach for the theoretical description of such a system. Let a film with a thickness d be made of a material with dielectric permittivity ε_2 be located between two semi-infinite dielectrics with dielectric permittivity ε_1 and ε_3 respectively.

s-polarization:



p-polarization:



The incidence of the *s*-polarized and *p*-polarized wave on the film

When a incident wave falls on the film, a reflected wave occurs in the medium ε_1 , a passing wave in the medium ε_3 and a superposition of waves going in different directions in the medium ε_2 . In *s*-polarization, the direction of the electric field is fixed, so we can proceed to consider scalar complex amplitudes:

$$\begin{cases} \tilde{E}_1 = \mathcal{E}e^{ik_{1x}x+ik_{1y}y} + r \cdot \mathcal{E}e^{-ik_{1x}x+ik_{1y}y} \\ \tilde{E}_2 = a \cdot \mathcal{E}e^{ik_{2x}x+ik_{2y}y} + b \cdot \mathcal{E}e^{-ik_{2x}x+ik_{2y}y} \\ \tilde{E}_3 = t \cdot \mathcal{E}e^{ik_{3x}(x-d)+ik_{3y}y} \end{cases}$$

In *p*-polarization, the direction of the magnetic field is fixed, so it is easier to proceed to the consideration of scalar complex amplitudes of the magnetic field:

$$\begin{cases} \tilde{B}_1 = \mathcal{B}e^{ik_{1x}x+ik_{1y}y} + r \cdot \mathcal{B}e^{-ik_{1x}x+ik_{1y}y} \\ \tilde{B}_2 = a \cdot \mathcal{B}e^{ik_{2x}x+ik_{2y}y} + b \cdot \mathcal{B}e^{-ik_{2x}x+ik_{2y}y} \\ \tilde{B}_3 = t \cdot \mathcal{B}e^{ik_{3x}(x-d)+ik_{3y}y} \end{cases}$$

A1 Show that $k_{1y} = k_{2y} = k_{3y}$.

A2 Find expressions for k_{1x} , k_{2x} and k_{3x} in terms of ω , c , θ_1 , ε_1 , ε_2 , ε_3 .

A3 For a wave with *s*-polarization (the electric field is directed along the surface), from the boundary conditions for the tangential components of the fields, obtain a system of linear equations for the coefficients r , A , b and t .

A4 Find the reflection coefficient. Bring the answer to the form:

$$r_s = \frac{A + Be^{2ik_{2x}d}}{AB + e^{2ik_{2x}d}}.$$

Express the real numbers A , B in terms of k_{1x} , k_{2x} and k_{3x} .

For convenience, use the variables $\kappa_{ix} = \frac{k_{ix}}{\varepsilon_i}$ for $i \in [1, 3]$.

A5 For a wave with p -polarization (the magnetic field is directed along the surface), from the boundary conditions for the tangential components of the fields, obtain a system of linear equations for the coefficients r , A , b and t .

A6 Find the reflection coefficient. Bring the answer to the form:

$$r_p = \frac{A + Be^{2ik_{2x}d}}{AB + e^{2ik_{2x}d}}.$$

Express the real numbers A , B in terms of k_{1x} , k_{2x} and k_{3x} .

We will perform preliminary calculations of the approximate shape of the ellipsometric spectrum of a thin layer of silicon dioxide on a silicon substrate. Until the end of Part A, we will neglect the optical dispersion, and assume $\varepsilon_{\text{Air}} = 1.00$, $\varepsilon_{\text{Si}} = 17.2 - 0.430i$, $\varepsilon_{\text{SiO}_2} = 2.13$. Note that silicon has an imaginary part of the dielectric constant responsible for absorption, due to which it is opaque. The use of complex permittivity does not change obtained expressions, if the trigonometric functions of the complex parameter are considered.

With the help of the A.py program, you can plot Ψ and Δ for different angles of incidence θ_1 and thicknesses d . At the input, the program receives the name Name of the series, the angle of incidence θ_1 from the air (in degrees) and the thickness d of the dioxide film in the form of a table written in the file A_in.txt .

There is an example of filling in a table in a file A_in.txt:

Name	$\theta_1, ^\circ$	d , nm
ex1	50	40
ex2	30	100
$\vdots$	$\vdots$	$\vdots$

The input data is limited to $20^\circ \leq \theta_1 \leq 80^\circ$, $d \leq 250$ nm. The name of the experiment should not contain spaces and tabs. The separator between the integer and fractional parts of numbers is a dot. The separator between the columns is a tab.

A7 Using the program A.py plot the dependence of Ψ and Δ on λ at different angles of incidence θ_1 and thicknesses d . Conduct a quantitative study and determine at which angle of incidence θ_{opt} the ellipsometric angle Ψ is most sensitive (the highest value $|\Delta\Psi/\Delta d|$) to changes in d in the range from 10 nm to 50 nm. Find the value of θ_{opt} with an accuracy of one degree. Since B3, use θ_{opt} as Φ to increase accuracy.

Part B. Kinetics of oxidation

In this part, the angle of incidence of the beam on the film will be indicated by Φ .

Together with a colleague, you are studying the deposition of thin oxide films on the surface of silicon during prolonged calcination in the dry air. Real silicon Si and silicon dioxide SiO₂ have optical dispersion, so the graphs Ψ, Δ from λ that will be offered to you in this part may slightly differ from those that you built in the part **A7**.

Attention! The optical dispersion of Si and SiO₂ is automatically embedded in all programs that you will work with in this part.

Your colleague was working late on the ellipsometer yesterday and forgot to write down the value of the angle Φ . He measured the thickness of the oxide film on two samples and transmitted the measurement results to you in files B1.txt and B2.txt.

B1 Using the program B1.py find by hand the value Φ and the thickness d of SiO_2 film. The program B1.py reads values from a file B1.txt automatically and outputs an approximating graph Ψ, Δ from λ , for the values of the angle Φ entered by you in degrees and the thickness d of the dioxide film in nanometers. The input data is limited: $20^\circ \leq \Phi \leq 80^\circ$, $d \leq 250$ nm.

B2 Using the program B2.py find by hand the value Φ and the thickness d of SiO_2 film. The program works similarly to the previous **B1**.

For a detailed study of the oxidation rate, you can get ellipsometric data from your colleague, measured from a real-time oxidizing silicon substrate. At the initial moment $t = 0$, he puts pure silicon inside the furnace with a temperature T , and every 2 min measures the spectrum Ψ, Δ from λ at a fixed angle Φ , which you tell him.

In the file B_in.txt you specify the name of the experiment, the angle Φ in degrees, the temperature T in degrees Celsius, and the duration of the experiment in minutes in the form of a table:

Name	$\Phi,^\circ$	$T,^\circ\text{C}$	$t, \text{ min}$
test	50	1000	40
$\vdots$	$\vdots$	$\vdots$	$\vdots$

The input data is limited as $20^\circ \leq \Phi \leq 80^\circ$, $800^\circ\text{C} \leq T \leq 1200^\circ\text{C}$, $t \leq 200$ min. The name of the experiment should **not** contain spaces and tabs. The separator between the integer and fractional parts of a number is a dot. The separator between the columns is a tab. Next, you run the program gen_data.exe which simulates your colleague conducting an experiment in accelerated mode. You can interrupt the execution of this program at any time, and the results of the measurements already performed will be saved on your computer. You can automatically process the measurement data using the program B.py which uses the same file B_in.txt. This program uses the results of the work gen_data.exe, selects the value of d for each point and calculates the statistical error of this value. Keep in mind that processing may take several minutes. The results of the program B.py are located in the /res folder in the form of tables and graphs $d(t)$. An example of an automatically generated program B.py tables in the /res folder:

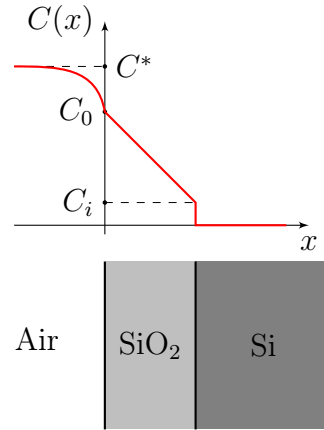
$t, \text{ min}$	$d, \text{ nm}$	$\Delta d, \text{ nm}$
2	17.3	0.3
$\vdots$	$\vdots$	$\vdots$

B3 Get the data for the dependence of the film thickness d of silicon dioxide on the time t during oxidation at a temperature of $t = 1000^\circ\text{C}$ for 120 min. Attach the graph obtained by the program B.py to the report.

Now let us consider the theoretical aspects of thermal oxidation. According to the Deal-Grove model, three processes are responsible for the rate of increase in the thickness of the silicon oxide layer:

1. diffusion of oxygen to the surface SiO_2 ;
2. diffusion of oxygen through the layer SiO_2 ;
3. oxidation reaction $\text{Si} + \text{O}_2 \rightarrow \text{SiO}_2$ at the interface SiO_2/Si .

Thus, the concentration of oxygen C (dimension mol/m^3) changes as it moves deeper into the sample. In the air, far away from the substrate, the oxygen concentration is C^* ; inside the silicon dioxide layer, the concentration decreases linearly from C_0 to C_i ; inside silicon, the oxygen concentration is zero.



Let us consider a model for describing these processes. Specific (i.e. per unit surface area) the flow rates of each of the three processes F_1 , F_2 , F_3 (units of measurement — $\text{mole}/(\text{c} \cdot \text{m}^2)$) we define as follows:

1. The diffusion rate is $F_1 = h(C^* - C_0)$, where h is a certain coefficient characterizing diffusion in the gas phase.
2. Counting the diffusion rate $F_2 = \frac{D}{d}(C_0 - C_i)$, where D is the diffusion coefficient of oxygen in the dioxide, depending on the temperature T , and d is the thickness of the silicon dioxide film.
3. The rate of the oxidation reaction $F_3 = kC_i$, where k is the constant of the rate of the oxidation reaction, depending on the temperature T .

B4	Express the growing velocity v of film thickness change in terms of the oxidation reaction rate F_3 , the molar mass μ_{SiO_2} and the density ρ_{SiO_2} of silicon dioxide. Assume that the air-dioxide boundary does not move.
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B5	Assuming the problem to be quasi-stationary (the establishment times of all processes are much less than the film rise time), express C_0 and C_i in terms of C^* , the velocity constants h , k , D and the film thickness d .
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B6	Get the differential equation for the thickness d of the dioxide film and get its solution with the initial condition $d(t_0) = d_0$ in the form $d^2 + Ad = B(t + \tau),$ where τ contains information about the initial conditions, and the constants A and B are related to the kinetics of oxidation. Express A and B in terms of h, k, D, C^*, μ_{SiO_2} and ρ_{SiO_2}.
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Experiments show that blowing with air (changing the profile $C(x)$ above the surface), doesn't change the rate of the process. Therefore we will consider $h \gg k$.

At the beginning of the oxidation process, the stationary model is not applicable, so it is used starting from the minimum thickness $d_0 = 25 \text{ nm}$. Enter into the program B7.py the data from the file B_in.txt and get the parameters B and B/A automatically. If the maximum film thickness obtained in the experiment is less than d_0 , then the program B7.py outputs "Final layer is too thin to approximate".

If the program returns the error "FileNotFoundError", run the files gen_data.exe and B.py sequentially with the input data you need, previously entered in the B_in.txt.

B7 Get the values of the parameters B and B/A for five different temperatures T . Save the results into a file B8_in.txt in the form of a table.

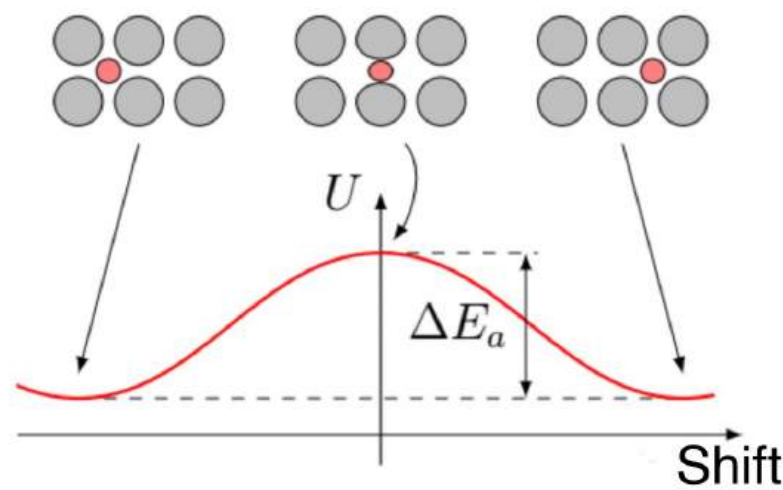
$T, ^\circ\text{C}$	$B, \text{nm}^2/\text{min}$	$\Delta B, \text{nm}^2/\text{min}$	$B/A, \text{nm}/\text{min}$	$\Delta(B/A), \text{nm}/\text{min}$
1000	174	4	1.21	0.02
$\vdots$	$\vdots$	$\vdots$	$\vdots$	$\vdots$

The dependence of the reaction rate constant k on temperature T is associated with the energy ΔE_b of breaking the bond Si – Si in a silicon crystal, and then the relation between them:

$$k \propto e^{-\Delta E_b/k_B T}.$$

The diffusion coefficient in solids is also related to temperature via the so-called activation energy ΔE_a , and the quantitative dependence has the form:

$$D \propto e^{-\Delta E_a/k_B T}$$



A qualitative diffusion scheme in a solid and the meaning of energy ΔE_a .

Using the program B8.py you can plot graphs $\ln(B \cdot \text{min}/\text{nm}^2)$ from $1/T$ and $\ln(B/A \cdot \text{min}/\text{nm})$ from $1/T$ and determine the parameters of the automatically drawn lines.

Boltzmann constant $k_B = 1.38 \cdot 10^{-23} \text{ J/C}$, elementary charge $e = 1.60 \cdot 10^{-19} \text{ C}$.

B8 Determine the values of ΔE_b and ΔE_a in electron volts. Attach the linearized graphs to the report.